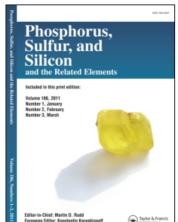
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Studies of Thiazolopyridines, Part 6: Synthesis and Antimicrobial Evaluation of Some Novel Thiazolo[3,2-a] Pyridine and Thiazolo[2',3':6,1]-pyrido[2,3-d]pyrimidine Derivatives

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Studies of Thiazolopyridines, Part 6: Synthesis and Antimicrobial Evaluation of Some Novel Thiazolo[3,2-a] Pyridine and Thiazolo[2',3':6,1]-pyrido[2,3-d]pyrimidine Derivatives

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Condensation of thiazolinone 1 with aromatic aldehydes yielded the corresponding methylidene derivatives 2a-f. Cyclization of compounds 2a-f with arylidenemalononitrile 3 (1:1 molar ratio) in ethanol in the presence of piperidine furnished the novel thiazolo[3,2-a]pyridines 5a-v, via Michael adduct 4. Compounds 5p,r were cyclized with malononitrile in the presence of piperidine to yield thiazolo[3,2-a][1,8]naphthryidines 7a,b. Thiazolo-[2',3':1,6]pyrido[2,3-d]pyrimidine 9a-c were obtained by cyclization of compounds 5c,p,r with formic acid. The structure of the synthesized compounds was established by analytical and spectral data. Also, some of the synthesized compounds were screened for antimicrobial activity in vitro.

Keywords Naphthryidine; thiazoline; thiazolo[3,2-a]pyridine;

Thiazolo[3,2-a]pyridines were reported to furnish various biological and pharmacological activities such as antimicrobial, bactericide, coronary dilator, antihypertensive, and muscle relaxant activities. In continuation with our work on the synthesis of thiazolo[3,2-a]pyridines, from readily available starting materials, 4-7 we report here on the synthesis of thiazolo[3,2-a] pyridine, thiazolo[3,2-a][1,8]naphthyridine and

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thiazolo[2′,3′:1,6]pyrido-[2,3-d]pyrimidine derivatives in order to investigate the antimicrobial activity of them.

RESULTS AND DISCUSSION

The starting material 1 was synthesized by cyclocondensation of cyanoacetamide with thioglycolic acid.8 Thiazoline 1 was condensed with various aromatic aldehydes in ethanol in the presence of piperidine under reflux to furnish 2-acetamido-5-arylmethylidene-4,5-dihydro-4thiazolinones 2a-f (Scheme 1). Cyclocondensation of compounds 2a-f with benzylidenemalononitriles 3 (1:1 molar ratio) in ethanol at reflux temperature in the presence of piperidine yielded the novel thiazolo [3,2a]-pyridines 5a-v in high yields (Scheme 1). The structures of all the compounds 5a-c were established using microanalysis and spectroscopy. The infrared spectra of thiazolopyridines 5a-v exhibited the presence of amino, cyano, and carbonyl (thiazolinone and carbamoyl) function groups. Also, the ¹H-NMR spectra revealed a signal characteristic for the 4H-pyridine. The formation of thiazolopyridine 5 is assumed to proceed via Michael addition of methylene function group in compound 2 to the benzylidene moiety 3 to yield Michael adduct 4 followed by intramolecular cyclization⁷ at the cyano group to form thiazolopyridine 5 (Scheme 1). Also, the structure of thiazolopyridine 5 was established by another synthetic route by ternary condensation of compound 2, aromatic aldehyde, and malononitrile (1:1:1 molar ratio) in ethanol in the presence of piperidine under reflux.

The reactivity of thiazolopyridine **5** towards malononitrile was studied. Thus, when compound **5p,r** were allowed to react with malononitrile in ethanol containing piperidine under reflux, the condensed pyridine derivatives **7a,b** were obtained. On the basis of analytical and spectral data, the other possible structure **6** was ruled out. The infrared spectrum of compound **7a** showed $\nu_{C=0}$ at 1698 cm⁻¹ characteristic for thiazolinone and $\nu_{C=0}$ at 1674 cm⁻¹ (CONH₂). Formation of **7** is assumed to proceed through the addition of an amino group in **3** to the cyano group of malononitrile followed by intramolecular cyclization to furnish **7** (Scheme 2).

Cyclization of compounds **5c**,**p**,**r** with formic acid furnished the novel thiazolo[2′,3′:6,1]pyrido[2,3-d]pyrimidines **9a–c**, via intermediate amide formation **9 8** followed by intramolecular cyclization with formic acid. Also, cyclocondensation of compound **5c** with acetic anhydride under reflux yielded the corresponding thiazolopyridopyrimidine derivative **10** (Scheme 3). Compound **10** was characterized through the absence of cyano functional group in its infrared spectrum.

2a; Ar= C_6H_4Cl-4 2b; Ar= $C_6H_4CH_3-4$ 2c; Ar= C_6H_4Cl-2 2d; Ar= C_6H_4OH-4 2e; Ar= $C_6H_4OCH_3-4$ 2f; Ar= C₆H₃(OCH₃)(OH)-3,4 5a; Ar= C_6H_4Cl-4 , Ar'= $C_6H_3(OCH_3)(OH)-3.4$ 5b; Ar= C_6H_4Cl-4 , Ar'= C_6H_4Cl-2 5c; $Ar = C_6H_4Cl-4$, $Ar' = C_6H_4OH-4$ 5d; $Ar = C_6H_4CH_3-4$, $Ar' = C_6H_4OH-4$ 5e; Ar= $C_6H_4CH_3$ -4, Ar'= C_6H_4Cl -2 5f; $Ar = C_6H_4CH_3-4$, $Ar' = C_6H_4Cl-4$ 5g; Ar= C_6H_4Cl -2, Ar'= C_6H_4OH -4 5h; Ar= C_6H_4Cl-2 , Ar'= $C_6H_3(OCH_3)(OH)-3,4$ 5i; Ar= C_6H_4Cl-2 , Ar'= C_6H_4Cl-4 5j; Ar= C₆H₄OH-4, Ar'= C₆H₄CH₃-4 5k; Ar= C_6H_4OH-4 , Ar'= C_6H_4Cl-2 5l; Ar= C₆H₄OH-4, Ar'= C₆H₄Cl-4 5m; Ar= C₆H₄OH-4, Ar'= C₆H₄OCH₃-4 5n; Ar= C_6H_4OH-4 , Ar'= $C_6H_3(OCH_3)(OH)-3,4$ **50**; Ar= $C_6H_4OCH_3$ -4, Ar'= $C_6H_3(OCH_3)(OH)$ -3,4 **5p**; Ar= $C_6H_4OCH_3$ -4, Ar'= C_6H_4Cl -2 5q; Ar= $C_6H_4OCH_3$ -4, Ar'= C_6H_4Cl -4 5r; $Ar = C_6H_4OCH_3-4$, $Ar' = C_6H_4CH_3-4$ 5s; Ar= C₆H₄OCH₃-4, Ar'= C₆H₄OH-4 5t; Ar= $C_6H_3(OCH_3)(OH)$ -3,4, Ar'= C_6H_4OH -4 5u; Ar= $C_6H_3(OCH_3)(OH)$ -3,4, Ar'= $C_6H_4CH_3$ -4 5v; Ar= $C_6H_3(OCH_3)(OH)$ -3,4, Ar'= C_6H_4Cl -2

ANTIMICROBAL ACTIVITY

SCHEME 1

Some of the synthesized compounds were screened in vitro for their antimicrobal activities against three strains: Staphylococcus aureus

$$(5p,r)$$

$$\begin{pmatrix}
O & Ar' & H & CN \\
H_2N - C & & & & & \\
H_2N - C & & & & & \\
NC & & & & & \\
NH_2 & & & & \\
NH_2 & & & & & \\
NH_2$$

7a; Ar= $C_6H_4OCH_3$ -4, Ar'= C_6H_4Cl -2
7b; Ar= $C_6H_4OCH_3$ -4, Ar'= $C_6H_4CH_3$ -4

SCHEME 2

9a; Ar= C_6H_4Cl -4 , Ar'= C_6H_4OH -4

9b; Ar= $C_6H_4OCH_3$ -4 , Ar'= C_6H_4Cl -2

9c; Ar= $C_6H_4OCH_3$ -4, Ar'= $C_6H_4CH_3$ -4

10; Ar= C_6H_4Cl -4 , Ar'= C_6H_4OH -4

SCHEME 3

TABLE I Antimicrobial Activity of the Synthesized Compounds and Inhibition Zones (mm). Standard: for Gram Positive and Gram Negative Bacteria, Ampicillin 25 μg mL $^{-1}$; for fungi: Mycostatine 30 μg mL $^{-1}$

Compound no.	Staphylococcus aureus (NCTC-7447)	Saracia maxima (ATCC-33910)	Aspergillus funigatus
5b	16	19	14
5d	12	12	18
5f	12	19	18
5h	14	17	14
5k	17	12	16
5p	19	18	15
5 q	15	18	17
5r	18	14	12
5s	20	20	25
5t	12	13	13
5 v	13	18	12
7a	20	21	20
9b	21	20	22

(NCTC-7447), Saracia maxima (ATCC-33910), and Aspergillus funigatus by the agar diffusion techniques. ¹⁰ The tested compounds were dissolved in N,N-dimethylformamide (DMF) to get a solution of $1000~\mu g$ mL⁻¹ concentration. The bacteria and fungi cultures were maintained on nutrient agar and Czapek's-Dox agar media, respectively. DMF showed no inhibition zones. The agar media were incubated with different microorganisms culture tested. After 24 h of incubation at 30°C for bacteria and 48 h of incubation at 28°C for fungi, the diameter of inhibition zone (mm) was measured (Table I). Ampicillin in a concentration $25~\mu g$ mL⁻¹ and mycostatine in a concentration $30~\mu g$ mL⁻¹ were used as references for antibacterial and antifungal activities, respectively. The results are illustrated in Table I. None of the tested compounds showed a superior activity than the reference.

EXPERIMENTAL

Melting points are uncorrected. IR spectra were recorded (KBr) on a Perkin Elmer 1650 spectrophotometer. 1H NMR spectra were recorded on a Varian Gemini spectrometer 200 (200 MHz), using DMSO-d₆ as a solvent and TMS as an internal standard. Chemical shifts are expressed as δ ppm units. Microanalytical data were obtained from the Microanalytical Data Unit at Cairo University. Physical data for the synthesized compounds are given in Table II. Also, the spectral data are collected in Table III.

TABLE II Characteristics Data for the Prepared Compounds

	TT: 1.1	G 1	2.5		Elemental analysis Calcd./Found %		
Compound no.	Yield (%)	Solvent cryst.	M.p. (°C)	Molecular formula (Mol. Wt.)		Н	N
2a	62	Benzene/ ethanol	240–2	$C_{12}H_9ClN_2O_2S$ (280)	51.42 51.40	3.21 3.00	10.00 10.10
2 b	57	Ethanol	197–8	$C_{13}H_{12}N_2O_2S$ (260)	60.10 60.05	4.61 4.60	10.76 10.70
2 c	71	Benzene	241-3	$C_{12}H_9ClN_2O_2S$ (280)	51.42 51.30	3.21 3.20	10.00 10.10
2d	83	Benzene/ ethanol	190–2	$C_{12}H_{10}N_2O_3S$ (262)	54.96 54.80	3.81 3.90	10.10 10.68 10.70
2e	62	Benzene	230–2	$C_{13}H_{12}N_2O_3S$ (276)	56.52 56.50	4.34 4.30	10.14 10.10
2f	56	Benzene	220–2	$C_{13}H_{13}N_2O_4S$ (293)	55.42 55.40	4.43 4.40	9.55 10.00
5a	63	Dioxane	256–8	$C_{23}H_{17}CIN_4O_4S$ (480.5)	57.44 57.40	3.52 3.50	11.65 11.60
5b	77	Benzene	258–9	$C_{22}H_{14}Cl_2N_4O_2S$ (469)	56.28 56.30	2.98 3.00	11.94 11.90
5 c	81	Benzene	286–8	$C_{22}H_{15}ClN_4O_3S$ (450.5)	58.60 58.60	$3.32 \\ 3.30$	12.43 12.40
5 d	69	Benzene	>300	$C_{23}H_{18}N_4O_3S$ (430)	64.18 64.20	$4.18 \\ 4.20$	13.02 13.10
5e	88	Benzene	220–2	$C_{23}H_{17}ClN_4O_2S$ (448.5)	61.53 61.50	$3.79 \\ 3.80$	12.48 12.50
5f	79	Benzene	226–8	$C_{23}H_{17}ClN_4O_2S$ (448.5)	$61.53 \\ 61.50$	3.79 3.80	12.48 12.40
5g	74	Benzene	271–3	$\substack{\text{C}_{22}\text{H}_{15}\text{ClN}_4\text{O}_3\text{S}\\ (450.5)}$	58.60 58.60	$3.32 \\ 3.30$	12.43 12.40
5h	64	Dioxane	262–4	$\substack{\text{C}_{23}\text{H}_{17}\text{ClN}_4\text{O}_4\text{S}\\ (480.5)}$	$57.44 \\ 57.40$	$3.52 \\ 3.50$	11.65 11.60
5i	66	Benzene	253–5	$\substack{ C_{22}H_{14}Cl_2N_4O_2S\\ (469)}$	56.28 56.30	$2.98 \\ 2.90$	11.94 11.90
5j	75	Benzene	>300	$C_{23}H_{18}N_4O_3S$ (430)	64.18 64.20	$4.18 \\ 4.20$	13.02 13.10
5k	80	Benzene/ ethanol	270–2	${ m C}_{22}{ m H}_{15}{ m ClN}_4{ m O}_3{ m S} \ (450.5)$	58.60 58.60	$3.32 \\ 3.30$	12.43 12.40
51	90	Dioxane	260–2	${ m C}_{22}{ m H}_{15}{ m ClN}_4{ m O}_3{ m S} \ (450.5)$	58.60 58.60	$3.32 \\ 3.30$	12.43 12.40
5m	72	Benzene	282–4	${^{\mathrm{C}_{23}\mathrm{H}_{18}\mathrm{N}_{4}\mathrm{O}_{4}\mathrm{S}}}_{(446)}$	61.88 61.70	$4.03 \\ 4.000$	12.55 12.50
5 n	82	Benzene	254–6	${ m C_{23}H_{18}N_4O_5S} \ (462)$	59.74 59.70	3.89 3.90	12.12 12.10
50	59	Benzene	262–4	$C_{24}H_{20}N_4O_5S$ (476)	60.50 60.40	$4.20 \\ 4.20$	11.76 11.80
5 p	77	Dioxane	190–2	$\substack{\text{C}_{23}\text{H}_{17}\text{ClN}_4\text{O}_3\text{S}\\ (464.5)}$	59.41 59.40 Continue	$3.65 \\ 3.50$	12.05 12.10

TABLE II	Characteristics	Data for	the Pr	epared Con	apounds
(Continue	d)				

Compound	Yield	Solvent	M.p. Molecular		Elemental analysis Calcd./Found %		•
no.	(%)	cryst.	(°C)	formula (Mol. Wt.)	С	Н	N
5q	67	Ethanol	160–2	$C_{23}H_{17}ClN_4O_3S$ (464.5)	59.41 59.40	3.65 3.40	12.05 12.00
5r	87	Dioxane	258–9	$C_{24}H_{20}N_4O_3S$ (444)	64.86 64.80	4.50 4.40	12.61 12.60
5s	67	Benzene	240–2	$C_{23}H_{18}N_4O_4S$ (446)	61.88 61.70	4.03 4.10	12.55 12.50
5t	72	Dioxane	>300	$C_{23}H_{18}N_4O_5S$ (462)	59.74 59.70	3.89 3.90	12.12 12.10
5u	68	Benzene/ ethanol	238–9	$C_{24}H_{20}N_4O_4S$ (460)	62.60 62.60	4.34 4.30	12.17 12.20
5v	85	Dioxane	270–2	$C_{23}H_{17}ClN_4O_4S$ (480.5)	57.44 57.40	3.53 3.50	11.65 11.60
7a	69	Benzene/ ethanol	288–9	$C_{26}H_{19}ClN_6O_3S$ (530.5)	58.81 58.80	3.58 3.60	15.83 15.80
7b	59	Benzene/ ethanol	218–9	$C_{27}H_{22}N_6O_3S$ (510)	63.52 63.50	4.31 4.20	16.74 16.70
9a	70	Dioxane	290	$C_{23}H_{15}ClN_4O_4S$ (478.5)	57.68 57.70	2.92 2.90	11.70 11.70
9b	81	Benzene	210–2	$C_{24}H_{17}ClN_4O_3S$ (492.5)	58.47 58.50	$3.45 \\ 3.40$	11.37 11.40
9c	63	Dioxane	>300	$C_{25}H_{20}N_4O_4S$ (472)	63.55 63.50	4.233 4.20	11.86 11.50
10	71	Dioxane	>300	$C_{24}H_{17}ClN_4O_4S$ (492.5)	58.48 58.50	3.45 3.60	11.37 11.30

2-(Acetamido-2-yl)-5-arylmethylidene-4,5-dihydro-4-thiazolinones (2a-f)

A mixture of compound 1 (0.01 mole), aromatic aldehyde (0.01 mole), and piperidine (0.5 mL) in ethanol (40 mL) was heated under reflux for 3 h. The product obtained was recrystallized from suitable solvent to give $\bf 2a-f$.

5-Amino-7-aryl-8-carbamoyl-3-oxo-2-arylmethylidene-2,3-dihydro-*H*-thiazolo[3,2-a]pyridine-6-carbonitriles (5a-v) *Method (A):*

A mixture of compound **2** (0.01 mole), benzylidene-malononitrile **3** (0.01 mole), and piperidine (0.5 mL) in ethanol (30 mL) was heated

TABLE III Spectral Data of the Synthesized Compounds

	•	-
Compound no.	IR/. $\nu_{ m max}~({ m cm}^{-1})$	¹ H NMR(δ/ppm)(DMSO-d ₆)
	max (cm)	11 11W11(0/ppiii)(DW50-46)
2a	3379, 3147 (NH ₂), 2977 (CH-aliph), 1720 (C=O; thiazolinone), 1666 (C=O, amide).	3.15 (s, 3H, CH ₂), 5.40 (s, 2H, NH ₂), 6.40–7.62 (m, 5H, Ar-H and methine-H).
2b	3334, 3127 (NH ₂), 2977 (CH-alph), 1710 (C=O; thiazolinone), 1658 (C=O; amide).	$\begin{array}{c} 2.20~(\mathrm{s},3\mathrm{H},\mathrm{CH}_3),3.10~(\mathrm{s},2\mathrm{H},\mathrm{CH}_2),\\ 5.20~(\mathrm{s},2\mathrm{H},\mathrm{NH}_2),6.30-7.70~(\mathrm{m},\\ 5\mathrm{H},\mathrm{Ar}\text{-H}~\mathrm{and}~\mathrm{methine}\text{-H}). \end{array}$
2e	3300, 3180 (NH ₂), 2980 (CH-aliph), 1712 (C=O; thiazolinone), 1660 (C=O; amide).	3.20 (s, 2H, CH ₂), 3.60 (s, 3H, OCH ₃), 5.30 (s, 2H, NH ₂), 6.25–7.82 (m, 5H, Ar-H and methine-H).
2f	3379, 3147 (NH ₂), 2930 (CH-aliph), 1720 (C=O; thiazolinone), 1666 (C=O; amide).	3.21 (s, 2H, CH ₂), 3.85 (s, 3H, OCH ₃), 5.76 (s, 2H, NH ₂), 6.90–7.46 (m, 4H, Ar-H and methine-H), 10.80 (s, 1H, OH).
5a	$3471, 3433 \text{ (NH}_2), 2180 \text{ ($C\!=\!N$)}, \\ 1710 \text{ ($C\!=\!O$; thiazolinone)}, \\ 1651 \text{ ($C\!=\!O$; amide)}.$	3.81 (s, 3H, OCH ₃), 3.99 (s, 1H, pyridine-H), 5.65 (s, 2H, NH ₂), 6.66–7.25 (m, 8H, Ar-H and methine-H), 7.39 (s, 3H, NH ₂), 10.10 (br, 1H, OH).
5b	3386, 3170 (NH ₂); 2183 (C≡N), 1697 (C=O; thiazolinone), 1651 (C=O; amide).	3.99 (s, 1H, pyridine-H), 5.76 (s, 2H, NH ₂), 6.73–7.50 (m, 9H, Ar-H and methine-H), 7.54 (s, 2H, NH ₂).
5 c	3433, 3332 (NH ₂); 2183 ($C \equiv N$); 1700 ($C = O$; thiazolinone), 1658 ($C = O$; amide).	4.08 (s, 1H, pyridine-H), 5.73 (s, 2H, NH ₂), 6.74–7.43 (m, 9H, Ar-H and methine-H), 7.47 (s, 1H, NH ₂), 10.90 (br, 1H, OH).
5d	3433, 3147 (NH ₂), 2109 ($\mathbb{C} = \mathbb{N}$); 1720 ($\mathbb{C} = \mathbb{O}$; thiazolinone), 1666 ($\mathbb{C} = \mathbb{O}$; amide).	2.19 (s, 3H, CH ₃), 4.15 (s, 1H, pyridine-H), 5.46 (s, 2H, NH ₂), 5.87–6.40 (m, 9H, Ar-H and methine-H), 6.42 (s, 2H, NH ₂), 10.30 (br, 1H, OH).
5e	3359, 3225 (NH ₂), 2191 ($\mathbb{C}\overline{=}\mathbb{N}$), 1681 ($\mathbb{C}\overline{=}0$; thiazolinone), 1651 ($\mathbb{C}\overline{=}0$; amide).	2.20 (s, 3H, CH ₃), 4.00 (s, 1H, pyridine-H), 5.10 (s, 2H, NH ₂), 6.74–7.51 (m, 9H, Ar-H and methine-H), 7.54 (s, 2H, NH ₂).
5g	$3433, 3320 \text{ (NH}_2), 2214 \text{ (C=N)}, \\ 1720 \text{ (C=O; thiazoline)}, \\ 1666 \text{ (C=O; amide)}.$	4.09 (s, 1H, pyridine-H), 5.73 (s, 2H, NH ₂), 6.74–7.42 (m, 9H, Ar-H and methine-H), 7.54 (s, 2H, NH ₂), 10.90 (br, 1H, OH).
5 h	3440, 3350 (NH ₂), 2191 ($\mathbb{C} = \mathbb{N}$), 1681 ($\mathbb{C} = \mathbb{O}$; thiazolinone), 1651 ($\mathbb{C} = \mathbb{O}$; amide).	3.90 (s, 3H, OCH ₃), 4.08 (s, 1H, pyridine-H), 5.70 (s, 2H, NH ₂), 6.75–7.48 (m, 8H, Ar-H and methine-H), 8.04 (s, 2H, NH ₂), 10.90 (br, 1H, OH). (Continued on next page)

TABLE III Spectral Data of the Synthesized Compounds (Continued)

Compound	I IR/. $v_{max}~({ m cm}^{-1})$	1 H NMR(δ /ppm)(DMSO- d_{6})
5i	3379, 3309 (NH ₂), 2190 (C≡N), 1720 (C=O; thiazolinone), 1666 (C=O; amide).	4.08 (s, 1H, pyridine-H), 5.70 (s, 2H, NH ₂), 6.82–7.33 (m, 9H, Ar-H and methine-H), 7.52 (s, 2H, NH ₂).
5k	3425, 3340 (NH ₂), 2180 (C≡N), 1700 (C=O; thiazolinone), 1666 (C=O; amide).	4.09 (s, 1H, pyridine-H), 5.73 (s, 2H, NH ₂), 6.79–7.52 (m, 9H, Ar-H and methine-H), 7.62 (s, 2H, NH ₂), 10.72 (br, 1H, OH).
51	3386, 3170 (NH ₂), 2183 (C≡N); 1690 (C=O; thiazolinone), 1658 (C=O; amide).	4.24 (s, 1H, pyridine-H), 5.70 (s, 2H, NH ₂), 6.74–7.58 (m, 9H, Ar-H and methine-H), 7.62 (s, 2H, NH ₂), 10.00 (br, 1H, OH).
5n	3433, 3355 (NH ₂), 2191 (C≡N), 1700 (C=O; thiazolinone), 1658 (C=O; amide).	3.74 (s, 3H, OCH ₃), 4.32 (s, 1H, pyridine-H), 5.71 (s, 2H, NH ₂), 6.73–7.46 (m, 8H, Ar-H and methine-H), 10.0 (br, 2H, 2OH).
50	3433, 3355 (NH ₂), 2214 (C≡N), 1700 (C=O; thiazolinone), 1651 (C=O; amide).	3.76, 3.89 (2s, 6H, 2OCH ₃), 4.07 (s, 1H, pyridine-H), 5.50 (s, 2H, NH ₂), 6.45–7.59 (m, 8H, Ar-H and methine-H), 8.03 (s, 2H, NH ₂), 11.20 (br, 1H, OH).
5p	3440, 3209 (NH ₂), 2214 (C≡N), 1700 (C=O; thiazolinone), 1651 (C=O; amide).	3.70 (s, 3H, OCH ₃), 4.12 (s, 1H, pyridine-H), 5.71 (s, 2H, NH ₂), 6.48–7.45 (m, 9H, Ar-H and methine-H), 7.51 (s, 2H, NH ₂).
5q	3355, 3232 (NH ₂), 2214 (C≡N), 1720 (C=O; thiazolinone), 1657 (C=O; amide).	3.74 (s, 3H, OCH ₃), 3.99 (s, 1H, pyridine-H), 5.47 (s, 2H, NH ₂), 6.03–7.42 (m, 9H, Ar-H and methine-H), 11.70 (s, 2H, NH ₂).
5 r	3463, 3348 (NH ₂), 2214 (C≡N), 1689 (C=O; thiazolinone), 1651 (C=O; amide).	2.2 (s, 3H, CH ₃), 3.67 (s, 3H, OCH ₃), 3.99 (s, 1H, pyridine-H), 5.67 (s, 2H, NH ₂), 6.68–7.42 (m, 9H, Ar-H and methine-H), 7.78 (s, 2H, NH ₂).
5s	3379, 3147 (NH ₂), 2191 (C≡N), 1720 (C=O; thiazolinone), 1666 (C=O; amide).	3.80 (s, 3H, OCH ₃), 4.11 (s, 1H, pyridine-H), 5.70 (s, 2H, NH ₂), 6.85–7.42 (m, 9H, Ar-H and methine-H), 7.41 (s, 2H, NH ₂), 10.56 (s, 1H, OH).
5t	3379, 3147 (NH ₂), 2221 (C≡N), 1720 (C=O; thiazolinone), 1658 (C=O; amide).	3.78 (s, 3H, OCH ₃), 4.09 (s, 1H, pyridine-H), 5.55 (s, 2H, NH ₂), 6.75–7.90 (m, 8H, Ar-H and methine-H), 7.90 (s, 2H, NH ₂), 9.01, 13.00 (2br, 2H, 2OH). (Continued on next page)

TABLE III Spectral Data of the Synthesized Compounds (Continued)

Compound no.	IR/. ν_{max} (cm ⁻¹)	¹ H NMR(δ/ppm)(DMSO-d ₆)
5v	3463, 3348 (NH ₂), 2214 (C≡N), 1689 (C≔O; thiazolinone), 1651 (C≔O; amide).	3.76 (s, 3H, OCH ₃), 4.09 (s, 1H, pyridine-H), 5.55 (s, 2H, NH ₂), 6.47–7.35 (m, 9H, Ar-H and methine-H), 7.90 (s, 2H, NH ₂), 11.30 (br. 1H, OH).
7a	3376, 3318 (NH ₂), 2190 ($\mathbb{C} = \mathbb{N}$), 1698 ($\mathbb{C} = \mathbb{O}$; thiazolinone), 1674 ($\mathbb{C} = \mathbb{O}$; amide).	3.60 (s, 3H, OCH ₃), 4.10 (s, 1H, pyridine-H), 5.62 (s, 2H, NH ₂), 6.48–7.45 (m, 9H, Ar-H and methine-H), 7.53, 8.62 (s, 2H, NH ₂).
7b	3318, 3200 (NH ₂), 2188 ($\mathbb{C} = \mathbb{N}$), 1700 ($\mathbb{C} = \mathbb{O}$; thiazolinone), 1676 ($\mathbb{C} = \mathbb{O}$; amide).	
9c	3358, 3176 (NH ₂), 1702 (C=O; thiazolinone), 1656 (C=O; amide).	2.30 (s, 3H, CH ₃), 3.69 (s, 3H, OCH ₃), 4.21 (s, 1H, pyridine-H), 5.67 (s, 2H, NH ₂), 6.75–7.50 (m, 9H, Ar-H and methine-H), 7.78 (s, 1H, NH), 8.90 (s, 1H, pyrimidine-H).
10	3318, 3200 (NH $_2$), 1700 (C=O; thiazolinone), 1676 (C=O; amide).	$\begin{array}{c} 3.10 \ (\mathrm{s}, 3\mathrm{H}, \mathrm{CH_3}), 4.32 \ (\mathrm{s}, 1\mathrm{H}, \\ \mathrm{pyridine-H}), 5.55 \ (\mathrm{s}, 2\mathrm{H}, \mathrm{NH_2}), \\ 6.74-7.43 \ (\mathrm{m}, 9\mathrm{H}, \mathrm{Ar-H} \ \mathrm{and} \\ \mathrm{methine-H}), 8.54 \ (\mathrm{s}, 1\mathrm{H}, \mathrm{NH}), \\ 10.72 \ (\mathrm{br}, 1\mathrm{H}, \mathrm{OH}). \end{array}$

under reflux for 6 h; the solid product which produced on heating was collected to give **5**.

Method (B):

A mixture of compound 2 (0.01 mole), aromatic aldehyde (0.01 mole), and malononitrile (0.01 mole) in ethanol (40 mL) was heated under reflux for 6 h; the solid product which produced on heating was collected to give $\bf 5$.

5-Aryl-8-arylmethylidene-6-carbamoyl-2,4-diamino-8,9-dihydro-*H*-thiazolo[3,2-a][1,8]naphthyridine-3-carbonitriles (7a,b)

General Procedure

A mixture of compound $\mathbf{5}$ (0.01 mole), malononitrile (0.01 mole) in ethanol (30 mL) was heated under reflux for 3 h. The solid product which precipitated upon heating was collected by filtration and recrystallized to give $\mathbf{7}$.

5-Aryl-8-arylmethylidene-4-oxo-3,4,8,9-tetrahydro-5*H*-thia-zolo-[2',3':1,6]pyrido[2,3-d]pyrimidine-6-carbamoyl (9a–c)

General Procedure

A mixture of compound **5** (0.01 mole) and formic acid (10 mL) was heated under reflux for 24 h. The reaction mixture was concentrated in vacuo and the precipitate was collected by filtration, washed with water, and recrystallized from a proper solvent to give **9**.

Formation of thiazolo[2',3':1,6]pyrido[2,3-d]pyrimidine derivative (10)

A solution of compound **5c** (0.01 mole) in acetic anhydride (5 mL) was refluxed for 24 h. The solid product, thus formed after cooling, was collected and recrystallized to give **10**.

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